Leveraging longer reference frames to characterize clumped isotope outliers and ¹⁷O effects

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Abstract

The stability of UW's reference frame minimizes it as an error source and allows unique questions to be addressed

• What controls the variance of carbonate and gas standards?

• Where do outliers come from and how do we account for them?

• What is the effect of ¹⁷O on temperature calibrations?

A stable reference frame across multiple years





• Heated gases (1000°C) and CO₂ equilibrated with water of varying isotopic composition at 4 and 60°C

• No discernable change over more than two years (January, 2015-July 2017)

• Allows other potential sources of Δ_{47} uncertainty to be characterized with less ambiguity

Reference frame gases from 2015 (triangle), 2016 (square) and 2017 (circle). Regressions are constructed from all data, but nearly identical lines would be generated from individual years and other subsets

Box and whisker plots of reference material in 6 month intervals. In all cases, the inner quartile (blue box) overlaps longterm mean of zero enrichment gases (n=2302) within 1 s.d.

Outliers Explained! predictable resetting of carbonate and gas standards



Standard deviation vs. mean Δ_{47} for ALL gas and carbonate reference materials

Motivation:

- When <u>ALL</u> carbonate and gas standards are considered minimum
- Increasing variance toward higher and lower values, except in uncleaned heated gases
- Suggests a predictable mechanism for outliers likely related to the preparatory line
- All are skewed toward $\Delta_{47} \sim 0.94\%$, except uncleaned 1000°C gases.
- When the 4σ standard deviation of zero gases (n=2302; bottom right, green vertical bars) is centered on the mode of all other populations, only intervals toward ~0.94% fall outside this range (blue bars).
- Two populations obvious in 1000°C cleaned, increasingly overlap at lower temperatures.

Implications and Open questions:

- Data further from room T (e.g. ETH1 and 2) may be noisier, even if they are homogenous. This is detectable only with the large sample sizes attainable with a long reference frame.
- Have you pitched an outlier? If yes, a similar process may affect your prep line. We would be interested in labs sharing ALL data in this project and other data compilations.
- What is the best way to cull outliers when measuring unknown samples fewer than dozens of times?
- What methods can further reduce water in preparatory systems to decrease the occurence of outliers?

Negligible effect of fluid ¹⁷O excess on Δ_{47} -T relationship in abiogenic calcites

• Evaporation produces waters with ¹⁷O excess of -14 and -83 per meg • Precipitate calcite at ~4, 22 and 45°C by mixing NaHCO₃ and CaCl₂ solutions (with carbonic anhydrase) and passively degasing • No significant difference from UW Kelson et al. (2017) calibration using local DI water with ¹⁷O excess of 35 per meg.

0.75

Summary and Conclusions

• UW reference frame is stable on

Explanation and Implications:

• Assuming a uniform offset between fluid and carbonate ¹⁷O excess, >100 per meg variations in solution values have small effect on K

> 1. K = ${}^{17}R_{std}({}^{18}R_{std})^{-\lambda}$, where $\lambda = 0.528$ 2. ¹⁷O excess = $\ln({}^{17}R/{}^{17}R_{std}) - 0.528 \cdot \ln({}^{18}R/{}^{18}R_{std})$ 3. K = exp(¹⁷O excess) • ${}^{17}R_{dtd}({}^{18}R_{dtd})^{-\lambda}$

¹⁷ O excess (per meg)	<u>K from eq 3</u>	$\underline{\Delta\Delta47}_{\text{Brand-varK}} (\% o)$
35	0.01022484	
0 (Brand)	0.01022461	
-14	0.01022434	-2.9 x 10 ⁻⁴
-83	0.01022364	8.9 x 10 ⁻⁶



multiannual timescales

• Outliers skew toward Δ_{47} equivalent to room T, with implications for low Δ_{47} standards.

• Outliers can be culled, but what approach is best?

• Fluid ¹⁷O excess does not detectably affect the Δ_{47} -T relationship over a range of >100 per meg

References

Kelson et al., 2017. Toward a universal carbonate clumped isotope calibration: Diverse synthesis and preparatory methods suggest a single temperature relationship. GCA 197, 104-131

Brand et al., 2010. Correction for the 17O interference in d13C measurements when analyzing CO2 with stable isotope mass spectrometry. Pure Appl. Chem 82, 8, 1719-1733